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4.8. The rate of the chemical reaction doubles for and increase of 10 K in absolute temperature from 298 K. Calculate E a. Ans. 4.9. The activation energy for the reaction, 2 HI(g) —-> H 2 +I 2 (g) is 209.5 k J mol -1 at 581 K. Calculate the fraction of molecules of reactants having energy equal to or greater than activation energy?

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