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Chapter 9 Compressible Flow 647 Solution: For CO₂, from Table A.4, take $k = 1.30$ and $R = 189 \text{ J/kg K}$. Compute the specific

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heat: $cp \ln\left(\frac{T_2}{T_1}\right) + 1.3(189)/(1.3 - 1) = 819 \text{ J/kg K}$. The results follow in sequence: (1) / (1.3 - 1) / 1. (a) $2 \ln\left(\frac{1}{2}\right) + 1(373 \text{ K}) \ln\left(\frac{80}{125}\right)$. (a) $kk. T_2 = 336 \text{ K}$ Ans

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